

## Article

# Optimization of a Novel Mandelamide-Derived Pyrrolopyrimidine Series of PERK Inhibitors

Michael E. Stokes <sup>1,\*</sup>, Matthew D. Surman <sup>2,\*</sup>, Veronica Calvo <sup>1</sup>, David Surguladze <sup>1</sup>, An-Hu Li <sup>1</sup>, Jennifer Gasperek <sup>3</sup>, Matthew Betzenhauser <sup>3</sup>, Guangyu Zhu <sup>3</sup>, Hongwen Du <sup>4</sup>, Alan C. Rigby <sup>1</sup> and Mark J. Mulvihill <sup>1</sup>

<sup>1</sup> HiberCell Inc., 619 West 54th Street, New York, NY 10019, USA

<sup>2</sup> Curia, 26 Corporate Circle, Albany, NY 12203, USA

<sup>3</sup> Curia, 1001 Main Street, Buffalo, NY 14203, USA

<sup>4</sup> Pharmaron Beijing Co., Ltd., No. 6, TaiHe Road, BDA, Beijing 100176, China

\* Correspondence: mstokes@hibercell.com (M.E.S.); matthew.surman@curiaglobal.com (M.D.S.)

† These authors contributed equally to this work.

**Abstract:** The protein kinase R (PKR)-like endoplasmic reticulum kinase (PERK) is one of three endoplasmic reticulum (ER) transmembrane sensors of the unfolded protein response (UPR) responsible for regulating protein synthesis and alleviating ER stress. PERK has been implicated in tumorigenesis, cancer cell survival as well metabolic diseases such as diabetes. The structure-based design and optimization of a novel mandelamide-derived pyrrolopyrimidine series of PERK inhibitors as described herein, resulted in the identification of compound **26**, a potent, selective, and orally bioavailable compound suitable for interrogating PERK pathway biology in vitro and in vivo, with pharmacokinetics suitable for once-a-day oral dosing in mice.

**Keywords:** ER stress; PERK; UPR; kinase; inhibitor; cancer; diabetes; small molecule; structure-activity-relationship (SAR)



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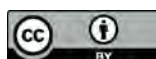
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## 1. Introduction

Cells must match the rates of protein synthesis with demand to avoid the onset of proteotoxic endoplasmic reticulum (ER) stress. In conditions of excess protein production, an accumulation of misfolded proteins in the ER lumen induces the unfolded protein response (UPR), which arrests global translation while enabling specific processes that restore protein homeostasis [1,2]. The UPR is mediated by three discrete signaling cascades, IRE1, ATF6, and EIF2AK3 (PERK) [3]. Upon activating the UPR, PERK phosphorylates eIF2a at Ser51, leading to widespread inhibition of protein translation with a few selected regulatory proteins translated to restore homeostasis and alleviate ER stress [1,3]. In cancer, acute activation of UPR enables tumors to overcome deleterious ER stress induced by rapid cell proliferation, nutrient deprivation, hypoxia, and drug treatments [4–6]. Pharmacological or genetic inhibition of PERK slows proliferation of tumor xenografts in mice [7,8], and small molecule PERK inhibitors have recently entered clinical trials for several oncology indications, including clear cell renal cell carcinoma (ccRCC) and multiple myeloma (NCT04834778; NCT05027594).

Acute ER stress and activation of UPR restores protein homeostasis and supports cell survival, whereas long-term ER stress shifts cells toward an apoptotic cell fate [9]. While intermittent ER stress occurs regularly in cell types that produce high abundance of protein, metabolic dysfunction can result in chronic ER stress that coincides with a number of metabolic and neurological diseases [9–12]. Type-1 Diabetes (T1D) is an example of a metabolic disease and autoimmune disorder characterized by PERK hyperactivation in insulin-producing  $\beta$ -cells of the pancreatic islets [11–13]. Immunohistochemistry (IHC) staining of pancreas sections from T1D patients revealed PERK pathway activation in

insulin-secreting beta cells, relative to healthy control patient samples [13], and PERK activation precedes the destructive autoimmune attack of beta cells in animal models of T1D [12,14]. Together, these findings highlight a potential role for PERK in diabetes and possible therapeutic benefit of PERK inhibitors beyond oncology.

While some cancers and metabolic diseases are characterized by PERK hyperactivation, genetic ablation of PERK results in glucose dysregulation and pancreatitis, suggesting that some basal level of PERK activity is essential to pancreatic health [15,16]. It is possible that the attenuation of PERK activity, rather than complete suppression, could provide therapeutic benefit while supporting normal pancreatic function. To enable future testing of this hypothesis, we sought to create a series of PERK inhibitors with stable PK profiles that could be administered at low doses to attenuate PERK yet tuned through higher doses to sufficiently inhibit PERK for tumor inhibition.

Previously described medicinal chemistry efforts around PERK inhibitors have made use of a variety of core hinge-binding regions, including aminopyridines, pyrrolopyrimidines, quinazolines, and quinolines [7,8,17]. Pyrrolopyrimidine cores have been developed and approved as JAK kinase inhibitors [18,19], and modifications to the core scaffold can have profound effects on stability and selectivity [17]. Here, we optimized the pyrrolopyrimidine core and leveraged extensive SAR information from our recently reported aminopyridine series of inhibitors to guide chemistry efforts aimed at optimizing the in vivo properties of pyrrolopyrimidine PERK inhibitors [7].

Our efforts were focused on developing potent and selective PERK inhibitors with suitable pharmacokinetic profiles to enable once or twice per day oral dosing in mouse models with a flat steady state exposure profile. The goal was to avoid large differences between  $C_{max}$  to  $C_{12h}$  or  $C_{24h}$  in order to achieve a constant and sustained pharmacodynamic response. This paper focuses on the structure-based design, structure–activity relationship (SAR), absorption, distribution, metabolism and excretion (ADME), and pharmacokinetic/pharmacodynamic (PK/PD) properties of a novel class of mandelamide-derived pyrrolopyrimidine PERK inhibitors leading to the identification of Compound **26**, with the desired flat steady state exposure profile. Compound **26** was characterized in a mouse pancreas assay and has been selected for further evaluation in animal models, which will be reported in due course.

## 2. Materials and Methods

### 2.1. Biochemical Assays

The inhibitory potencies of **26** inhibition against PKR, HRI, PERK and GCN2 were evaluated in cell-free biochemical assays following manufacturer’s instructions (Carna; Table 1). In brief, **26** was dissolved in DMSO and serially diluted 3-fold to generate concentrations ranging from 10  $\mu$ M to 0.2 nM in a 384-well plate using a TECAN EVO200 automated liquid handler. 10 nL stock solution was transferred into each well of a 384well assay plate by an Echo550 and an equivalent volume of DMSO was added to the control wells. To each assay well, recombinant protein was suspended in 20  $\mu$ L reaction buffer (listed below) and incubated at 25 °C for 1.5 h while shaking at 1250 rpm. Following incubation, 10  $\mu$ L of a FRET dilution buffer including 20 mM EDTA and 4 nM Tb-anti-peIF2a [pSer51] (Invitrogen, Cat# PV4815) was added to each well and incubated at 25 °C for 2 h while shaking at 600 rpm in the dark. Analysis was performed using an Envision Microplate Reader (PerkinElmer).

**Table 1.** Reagents and conditions used in biochemical characterization of ISR kinases.

Protein	Carna Cat#	Assay Buffer
GCN2	05-153	GCN2 (1 nM); DTT (2 mM); ATP (150 $\mu$ M); GFP-eIF2a (100 nM; Invitrogen Cat# PV4809)
HRI	05-154	HRI (0.12 nM); DTT (2 mM); ATP (30 $\mu$ M); GFP-eIF2a (100 nM; Invitrogen Cat# PV4809)
PERK	05-155	PERK (0.09 nM); DTT (2 mM); ATP (8 $\mu$ M); GFP-eIF2a (100 nM; Invitrogen Cat# PV4809)
PKR	05-156	PKR (0.11 nM); DTT (2 mM); ATP (6 $\mu$ M); GFP-eIF2a (100 nM; Invitrogen Cat# PV4809)

## 2.2. PERK Crystallization and Structure Determination

Human PERK (575-1094  $\Delta$ 670-874) was purified as described previously [7]. Purified PERK protein at 11 mg/mL was mixed with 10 mM compound **11**, **24**, or **26** (in DMSO) to a final protein-inhibitor molar ratio 1:2. The PERK-inhibitor mixture was incubated on ice for 2 h before crystallization. The crystals with compound **11** or **26** were grown at 20 °C in sitting drops by combining 2.0  $\mu$ L PERK-inhibitor mixture, 2.0  $\mu$ L reservoir solution (12–14% PEG3350, 4% tacsimate pH7.0), and 0.4  $\mu$ L seed stock which was equilibrated over a 500  $\mu$ L reservoir solution. The crystal with compound **24** was grown at 4 °C in a hanging drop by combining 1.5  $\mu$ L PERK-inhibitor mixture, 1.5  $\mu$ L reservoir solution (9% PEG3350, 180 mM Na/K tartrate, 100 mM HEPES pH 7.0) which was equilibrated over a 500  $\mu$ L reservoir solution. The crystals were grown to 0.2–0.5 mm over a three-week period before harvesting for analysis. The crystal was transferred stepwise to a cryo-solution with the crystallization cocktail plus 20% glycerol before being frozen in liquid nitrogen. Diffraction data were collected at the IMCA-CAT beamline 17ID for compound **11** or GMCA-CAT beamline 23IDD for compound **24** or **26** at the Advanced Photon Source at Argonne National Laboratory using a Pilatus 6M detector. The diffraction images were processed with DIALS [20] and scaled with AIMLESS [21]. The structure was solved by molecular replacement using PDB 4X7J [17] as starting model by Phaser [22]. The structure was manually built using Coot [23] and subsequently refined using Refmac5 [24]. The crystallographic figures were generated by CCP4MG [25], and the statistics of data collection and refinement are summarized in Supplemental Table S1.

## 2.3. Pharmacokinetic and Pharmacodynamic Analysis in Plasma and Pancreas

For PK analysis of pyrrolopyrimidine analogs in mouse, rat, and dog plasma, the oral formulation was prepared as a solution containing 20% (*w/v*) Captisol in 25 mM NaH<sub>2</sub>PO<sub>4</sub> buffer (pH 2). Compounds were administered orally by gavage at 10 mg/kg to female CD1 mice. After dosing, plasma samples were collected to characterize the PK profiles at 0.25, 0.5, 1, 2, 4, 8, 12 and 24 h post-dose. The plasma concentration of compound was determined by protein precipitation with acetonitrile and liquid chromatography with mass spectrometric detection (LC-MS/MS). Parameters were estimated using Phoenix (WinNonlin) pharmacokinetic software version 6.1.0 using a non-compartmental approach consistent with the oral route of administration. PK analysis of **26** in plasma and pancreas followed a similar analytical method as above, with minor modifications. Compound **26** was suspended in a vehicle consisting of 0.5% methylcellulose (400 cP) and 0.1% Tween80 in water and administered to BALB/c nude mice by oral gavage at 0.3, 1, 3, 10, 30 mg/kg. Plasma and pancreas were sampled from 5 mice per group following a single oral administration at 1, 4, 8, 12, 24 h post-dose.

Total protein was extracted from pancreas tissues sampled during PK studies described above. Frozen pancreas tissue was homogenized by polytron in lysis buffer consisting of 2x Laemmli SDS sample buffer (Novex), supplemented with 10% BME (Gibco), 1X benzonase (EMD Millipore Sigma), phosphatase inhibitors (Sigma) and Mini protease inhibitor tablet (Roche). Homogenate was incubated at room temperature for 10', then boiled for 10', followed by centrifugation for 10' at max speed on a benchtop centrifuge. Protein analysis performed using JESS high-throughput protein analysis instrument (Biotechne) using antibodies for phosphoPERK developed internally and PERK (CST).

## 2.4. Biochemical and Cell-Based Characterization of Pyrrolopyrimidine Analogs

Methods used to evaluate biochemical PERK activity are described and published elsewhere [7]. In brief, biochemical PERK activity is evaluated using a LanthaScreen™ (PerkinElmer) TR-FRET assay to detect phosphorylation of a GFP-tagged eIF2a substrate (ThermoFisher). Excitation at 340 nm of a terbium chelate donor fluorophore in an eIF2a antibody results in energy transfer to GFP acceptor fluorophore upon eIF2a phosphorylation by PERK. Reaction buffer was composed of 50 mM HEPES (pH 7.4), 10 mM MgCl<sub>2</sub>, 1 mM EGTA and 0.01% (*v/v*) Brij™-35. Reactions were initiated by the addition of substrate and

ATP, followed by incubation for 1 h at RT prior to addition of EDTA and the anti-peIF2a antibody. FRET was measured by an EnVision Plate Reader (PerkinElmer). Data were plotted as percent inhibition along a 10-point 3-fold dilution series of inhibitors. IC<sub>50</sub> values were calculated using 4-parameter logistical fitting in XLFit.

Methods used to evaluate in vitro cellular PERK inhibition are described and published elsewhere [7]. In brief, stable HEK293 cells were created using lentiviral particles harboring a GFP-eIF2a expression vector. Transfectants were selected by puromycin and enriched using fluorescent cell sorting based on GFP expression. HEK293-EGFP-eIF2a cells were plated at 5000 cells/well in 384-well assay plates and incubated overnight. The following day, pyrrolopyrimidine analogs were added to wells using an Echo acoustic dispenser and incubated for 30 min. ER stress was induced by the addition of tunicamycin to the wells at 1 mM for 2 h. Cells were lysed and peIF2a was evaluated using a TR-FRET system, as described above.

### 2.5. In Vivo Tumor Xenograft Studies

In vivo studies of the effect of pyrrolopyrimidines on 786-O xenograft inhibition follow a method published elsewhere [7]. Female BALB/c nude mice were inoculated subcutaneously with 786-O tumor cells ( $5 \times 10^6$ ) in 0.1 mL of PBS. Animals were randomized when tumors reached 100–200 mm<sup>3</sup> into treatment groups. Pyrrolopyrimidine PERK inhibitors **24**, **26** or **70** were dissolved in vehicle solution consisting of 20% (*w/v*) Captisol in 25 mM NaPO<sub>4</sub> buffer (pH2) and administered twice daily (BID) by oral gavage for 29 days. Tumor volumes were measured by caliper and the volume was expressed in mm<sup>3</sup> using the formula:  $V = (L \times W \times W)/2$ , where V, L, and W represent tumor volume, length, and width, respectively. All studies were conducted following an IACUC-approved protocol (AN-1903-05-1798). Experimental data management and reporting were in accordance with applicable Crown Bioscience's Guidelines and Standard Operating Procedures.

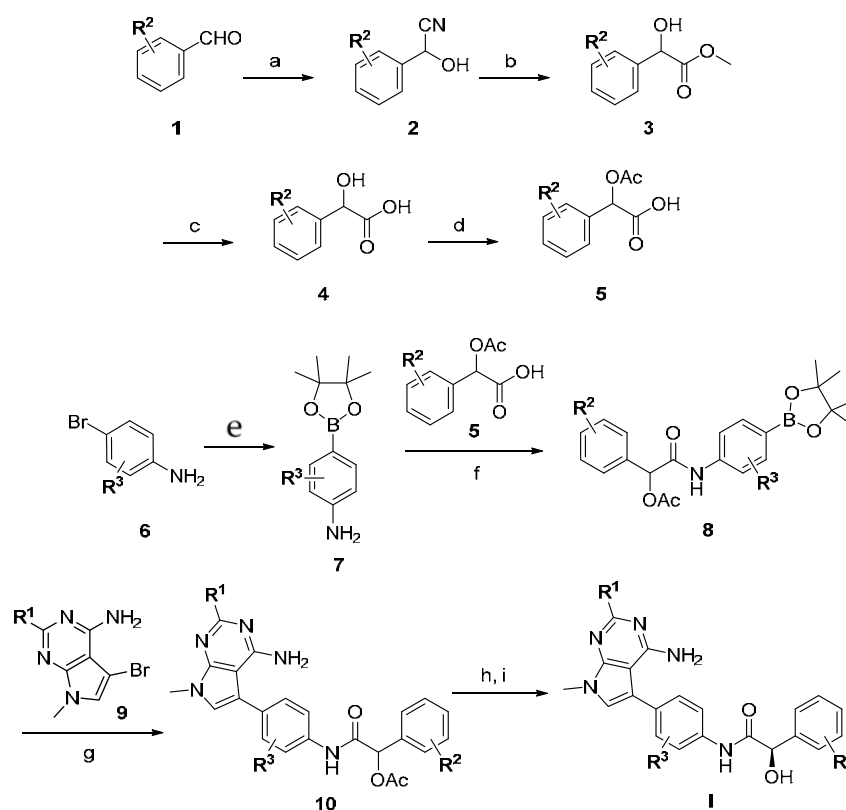
## 3. Results and Discussion

Small molecule PERK inhibitors across a variety of chemotypes have been previously described in the literature and captured in an earlier publication from our group around an aminopyridine series of PERK inhibition [7]. Herein, we report the synthesis and discovery of a series of novel mandelamide-derived pyrrolopyrimidines (**I**) that are highly potent, selective, and orally bioavailable PERK inhibitors. These medicinal chemistry efforts were iteratively supported by an X-ray crystallography structure-based approach to develop and refine SAR information. Collectively this approach permitted us to optimize the physiochemical and ADME/PK properties of this chemical series. Different R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, Ar and L groups have been introduced into our PERK small molecule series to tune the potency, selectivity and drug-like properties, such as aqueous solubility and cellular permeability.

The general synthesis for our pyrrolopyrimidine PERK inhibitors is shown in Scheme 1. The terminal ring substitution with different R<sup>2</sup> groups was introduced through the corresponding aldehyde **1**. Cyanohydrin **2** formation with TMS-cyanide and zinc(II) iodide followed by acid hydrolysis of the nitrile in methanol provided methyl mandelic ester **3**. Hydrolysis of the ester with lithium hydroxide and acylation of the benzylic alcohol gave the protected mandelic acid **5**.

The proximal ring substitution with different R<sup>3</sup> groups was introduced through bromo aniline **6**. Palladium-mediated borylation of **6** to give the aniline boronate ester **7** was followed by T<sub>3</sub>P coupling with acid **5** to provide amide **8**. Boronate ester **8** underwent Suzuki coupling with pyrrolopyrimidine bromide **9** to give racemic acetyl-protected product **10**. Deacylation under basic conditions followed by chiral separation of the enantiomers provided the final product **I**.





**Scheme 1.** General synthesis of PERK inhibitors. Reaction conditions: (a) TMSCN (1.2 equiv.), ZnI<sub>2</sub> (0.1 equiv.), 0 °C; (b) 3M HCl in MeOH, 0 °C to rt, 60–80% over two steps; (c) LiOH·H<sub>2</sub>O (1.2 equiv.), THF, H<sub>2</sub>O, MeOH, rt, 50–85%; (d) AcCl (2 vol.), 0 °C to rt, 70–85%; (e) B<sub>2</sub>pin<sub>2</sub> (1.2 equiv.), KOAc (3 equiv.), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (5 mol%), NEt<sub>3</sub> (10 vol%), 90 °C; (f) 5 (1.2 equiv.), KOAc (3 equiv.), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (5 mol%), NEt<sub>3</sub> (10 vol%), 90 °C; (g) 9 (1.2 equiv.), KOAc (3 equiv.), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (5 mol%), NEt<sub>3</sub> (10 vol%), 90 °C; (h) DIPEA (3 equiv.), 50% propylphosphonic anhydride (1.5 equiv.), EtOAc (10 vol%), CH<sub>2</sub>Cl<sub>2</sub>, 28 °C, 25–70%; (i) K<sub>2</sub>CO<sub>3</sub> (2 equiv.), MeOH, rt, 80–90%; (j) Chiral SFC separation, 80 °C, 25–70%; (h) K<sub>2</sub>CO<sub>3</sub> (2 equiv.), MeOH, rt, 80–90%; (i) Chiral SFC separation.

The proximal ring substitution with different R<sup>3</sup> groups was introduced through the bromo-aniline **6**. Palladium-mediated boronation of **6** to give the amine boronate ester **7** was followed by Suzuki coupling with acid **5** to provide amide **8**. Boronate ester **8** underwent Suzuki coupling with pyrrolopyrimidine bromide **9** to give racemic acetyl-protected product **10**. Deacylation under basic conditions followed by chiral separation of the enantiomers provided the final product **I**.

The PERK activity of analog **I** was evaluated using cell-free biochemical assays. In these assays, PERK phosphorylation of eIF2 $\alpha$  was assessed using a FRET-based labelled protein assay [7]. Compounds were then evaluated for cellular activity by quantifying inhibition of eIF2 $\alpha$  phosphorylation in HEK293 cells treated with the ER inducer tunicamycin [7]. Together, these data read out the potency of a given analog against PERK as well as the cellular permeability to enable bioactivity.

Initial SAR studies in the pyrrolopyrimidine series focused on the linker (L) region between the central amide and terminal aryl ring (Table 2). Hydroxyl substitution on this group was found to increase potency by 2–3-fold (**13** vs. **14**) with preference for the (R)-over the (S)-enantiomer (**11** vs. **12**). Expansion or contraction of the length of the linker **L** was not tolerated, resulting in at least 20-fold losses in PERK biochemical potency and nearly 50- to 100-fold losses in cellular potency (**14** vs. **15** and **16**, respectively). Difluoro substitution of the linker also caused a substantial loss in potency, as did replacement of the CH(OH) group with an O, NH, or NCH<sub>3</sub> group (**13** vs. **17**, **18**, **19**, and **20**, respectively). Seven-fold losses in both biochemical and cellular potency were observed when the hydroxyl group was converted to the methyl ether (**13** vs. **21**). Even more substantial losses were associated with replacement of the hydroxyl with an amino group (**13** vs. **23**). Addition of a methyl group to the linker to form a tertiary alcohol also caused a substantial loss in potency and nearly 50- to 100-fold losses in cellular potency (**14** vs. **15** and **16**, respectively). Difluoro substitution of the linker also caused a substantial loss in potency, as did replacement of the CH(OH) group with an O, NH, or NCH<sub>3</sub> group (**13** vs. **17**, **18**, **19**, and **20**, respectively).

Seven-fold losses in both biochemical and cellular potency were observed when the

Table 1. In vitro data of ERK5 inhibitors.		IC <sub>50</sub> (nM)	IC <sub>50</sub> (nM)
1	R <sup>1</sup>	0.0022	0.0024
2	R <sup>1</sup>	0.0022	0.0024
3	R <sup>1</sup>	0.0022	0.0024
4	R <sup>1</sup>	0.0022	0.0024
5	R <sup>1</sup>	0.0022	0.0024
6	R <sup>1</sup>	0.0022	0.0024
7	R <sup>1</sup>	0.0022	0.0024
8	R <sup>1</sup>	0.0022	0.0024
9	R <sup>1</sup>	0.0022	0.0024
10	R <sup>1</sup>	0.0022	0.0024
11	R <sup>1</sup>	0.0022	0.0024
12	R <sup>1</sup>	0.0022	0.0024
13	R <sup>1</sup>	0.0022	0.0024
14	R <sup>1</sup>	0.0022	0.0024
15	R <sup>1</sup>	0.0022	0.0024
16	R <sup>1</sup>	0.0022	0.0024
17	R <sup>1</sup>	0.0022	0.0024
18	R <sup>1</sup>	0.0022	0.0024
19	R <sup>1</sup>	0.0022	0.0024
20	R <sup>1</sup>	0.0022	0.0024
21	R <sup>1</sup>	0.0022	0.0024
22	R <sup>1</sup>	0.0022	0.0024
23	R <sup>1</sup>	0.0022	0.0024
24	R <sup>1</sup>	0.0022	0.0024
25	R <sup>1</sup>	0.0022	0.0024
26	R <sup>1</sup>	0.0022	0.0024
27	R <sup>1</sup>	0.0022	0.0024
28	R <sup>1</sup>	0.0022	0.0024
29	R <sup>1</sup>	0.0022	0.0024
30	R <sup>1</sup>	0.0022	0.0024
31	R <sup>1</sup>	0.0022	0.0024
32	R <sup>1</sup>	0.0022	0.0024
33	R <sup>1</sup>	0.0022	0.0024
34	R <sup>1</sup>	0.0022	0.0024
35	R <sup>1</sup>	0.0022	0.0024
36	R <sup>1</sup>	0.0022	0.0024
37	R <sup>1</sup>	0.0022	0.0024
38	R <sup>1</sup>	0.0022	0.0024
39	R <sup>1</sup>	0.0022	0.0024
40	R <sup>1</sup>	0.0022	0.0024
41	R <sup>1</sup>	0.0022	0.0024
42	R <sup>1</sup>	0.0022	0.0024
43	R <sup>1</sup>	0.0022	0.0024
44	R <sup>1</sup>	0.0022	0.0024
45	R <sup>1</sup>	0.0022	0.0024
46	R <sup>1</sup>	0.0022	0.0024
47	R <sup>1</sup>	0.0022	0.0024
48	R <sup>1</sup>	0.0022	0.0024
49	R <sup>1</sup>	0.0022	0.0024
50	R <sup>1</sup>	0.0022	0.0024
51	R <sup>1</sup>	0.0022	0.0024
52	R <sup>1</sup>	0.0022	0.0024
53	R <sup>1</sup>	0.0022	0.0024
54	R <sup>1</sup>	0.0022	0.0024
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56	R <sup>1</sup>	0.0022	0.0024
57	R <sup>1</sup>	0.0022	0.0024
58	R <sup>1</sup>	0.0022	0.0024
59	R <sup>1</sup>	0.0022	0.0024
60	R <sup>1</sup>	0.0022	0.0024
61	R <sup>1</sup>	0.0022	0.0024
62	R <sup>1</sup>	0.0022	0.0024
63	R <sup>1</sup>	0.0022	0.0024
64	R <sup>1</sup>	0.0022	0.0024
65	R <sup>1</sup>	0.0022	0.0024
66	R <sup>1</sup>	0.0022	0.0024
67	R <sup>1</sup>	0.0022	0.0024
68	R <sup>1</sup>	0.0022	0.0024
69	R <sup>1</sup>	0.0022	0.0024
70	R <sup>1</sup>	0.0022	0.0024
71	R <sup>1</sup>	0.0022	0.0024
72	R <sup>1</sup>	0.0022	0.0024
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75	R <sup>1</sup>	0.0022	0.0024
76	R <sup>1</sup>	0.0022	0.0024
77	R <sup>1</sup>	0.0022	0.0024
78	R <sup>1</sup>	0.0022	0.0024
79	R <sup>1</sup>	0.0022	0.0024
80	R <sup>1</sup>	0.0022	0.0024
81	R <sup>1</sup>	0.0022	0.0024
82	R <sup>1</sup>	0.0022	0.0024
83	R <sup>1</sup>	0.0022	0.0024
84	R <sup>1</sup>	0.0022	0.0024
85	R <sup>1</sup>	0.0022	0.0024
86	R <sup>1</sup>	0.0022	0.0024
87	R <sup>1</sup>	0.0022	0.0024
88	R <sup>1</sup>	0.0022	0.0024
89	R <sup>1</sup>	0.0022	0.0024
90	R <sup>1</sup>	0.0022	0.0024
91	R <sup>1</sup>	0.0022	0.0024
92	R <sup>1</sup>	0.0022	0.0024
93	R <sup>1</sup>	0.0022	0.0024
94	R <sup>1</sup>	0.0022	0.0024
95	R <sup>1</sup>	0.0022	0.0024
96	R <sup>1</sup>	0.0022	0.0024
97	R <sup>1</sup>	0.0022	0.0024
98	R <sup>1</sup>	0.0022	0.0024
99	R <sup>1</sup>	0.0022	0.0024
100	R <sup>1</sup>	0.0022	0.0024

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hydroxyl group was converted to a methyl ether (13 vs. 21). Even more substantial losses were associated with replacement of the hydroxyl with an amino group (15 vs. 23). Addition of a methyl group to the linker to form a tertiary alcohol also caused a substantial loss in potency (13 vs. 22). Based on these findings, it was concluded that (R)-1 is the preferred L group.

Table 2. In vitro data of 4-ERK inhibitors.  
Table 2. Cont.

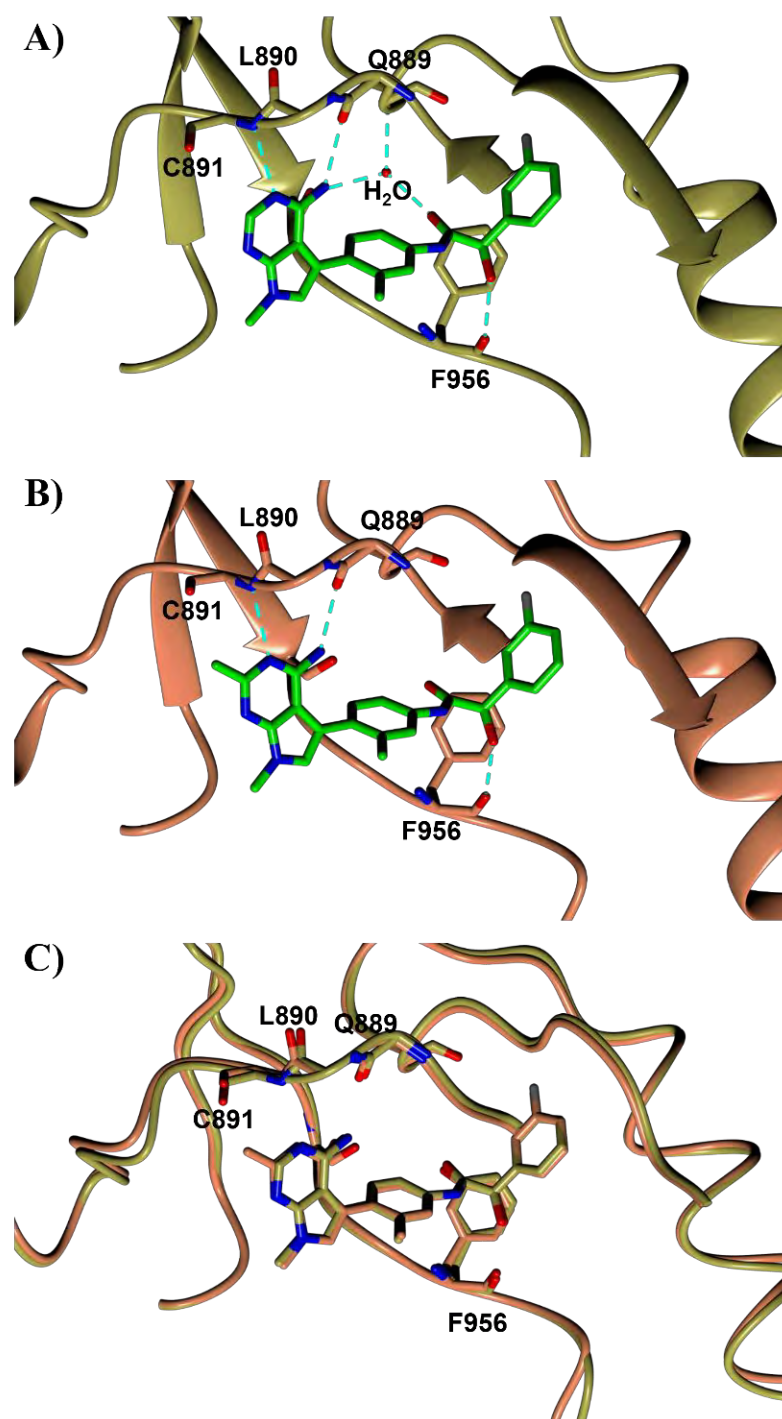
Compd	R <sup>1</sup>	IC <sub>50</sub> (nM)	IC <sub>50</sub> (nM)
25	Me	0.0028	0.005
26	Me	0.0028	0.015
27	Me	0.043	0.05
28	Me	0.0018	0.016
29	Me	0.0018	0.016
30	Me	0.0078	0.016
31	Me	0.107	0.713
32	Me	0.107	0.713
33	Me	0.007	0.713
34	Me	0.014	0.083
35	Me	0.014	0.083
36	Me	0.014	0.083
37	Me	0.022	0.084
38	Me	0.0059	0.042
39	Me	0.0059	0.042
40	Me	0.0059	0.042
41	Me	0.0078	0.057
42	Me	0.0059	0.042
43	Me	0.0059	0.042
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Previously reported SAR of PERK inhibitors have revealed modifications to the hinge binding region that have profound effects on binding affinity to PERK as well as selectivity versus the kinome [7,17]. Structure based design modelling was used to focus efforts on optimizing the pyrrolopyrimidine hinge-binding region of the molecule. Inhibitor **11** was co-crystallized with PERK into a trigonal space group P3<sub>2</sub>21 with comparable cell parameters as PDB 4X7O. Compound **11** formed three hydrogen bonds directly with the carbonyl of Q889, the amide of C891, and the carbonyl of F956 from PERK, respectively (Figure 1A). In addition, compound **11** forms two hydrogen bonds with a water molecule, and this molecule hydrogen bonds to the amide of V652. In the ATP binding hinge region, substitution at the C2-position (R1) of the pyrrolopyrimidine ring appeared feasible, filling a small pocket near C891. A methyl group located at the R1 position of compound **11** afforded analog **24** whose co-crystal structure with PERK confirmed the occupation of the small pocket near C891 along with the close contact of the C2-methyl group with the carbonyl of C891 with a distance of 3.14 Å (Figure 1B). Although a weak CH...O hydrogen bond is feasible, this contact appears to be more repulsive. As a result, a small rotation of the main chain amino acid Q889 and L890 of PERK, along with a minor shift of the pyrrolopyrimidine ring of inhibitor **24** are observed compared to those in the structure of **11** (Figure 1C). These shifts maintain the hydrogen bonds with similar distances. Due to a lower resolution (2.92Å) of the compound **24** structure (Table S1), the aforementioned water molecule in the compound **11** structure was not observed. This water molecule is assumed to locate at the same position, as all hydrogen bond donor and acceptor atoms are still presented at similar positions in the compound **24** structure. These observations indicate that the reduction of potency of **24** is likely caused by repulsion from the methyl group rather than the disruption of the hydrogen bonding pattern.

While having a minor effect on PERK potency, the methyl substituent on the pyrrolopyrimidine ring had a profound effect on the kinase selectivity. Compounds **11** and **24** were screened in the DiscoverX scanMAX<sup>SM</sup> panel of 468 kinases [26]. The resulting TREEspot<sup>TM</sup> interaction maps (Figure 2) revealed that the methyl substituent in **24** led to superior selectivity versus **11**. The S(35), S(10), and S(1) selectivity scores at a concentration of 1 µM were 0.005, 0, and 0 for **24** versus 0.072, 0.035, and 0 for **11**. Given the superior selectivity of **24** but slight loss in potency versus **11**, further optimization of the series was carried out with the 2-methyl substitution on the pyrrolopyrimidine ring to further improve potency and drug-like properties.

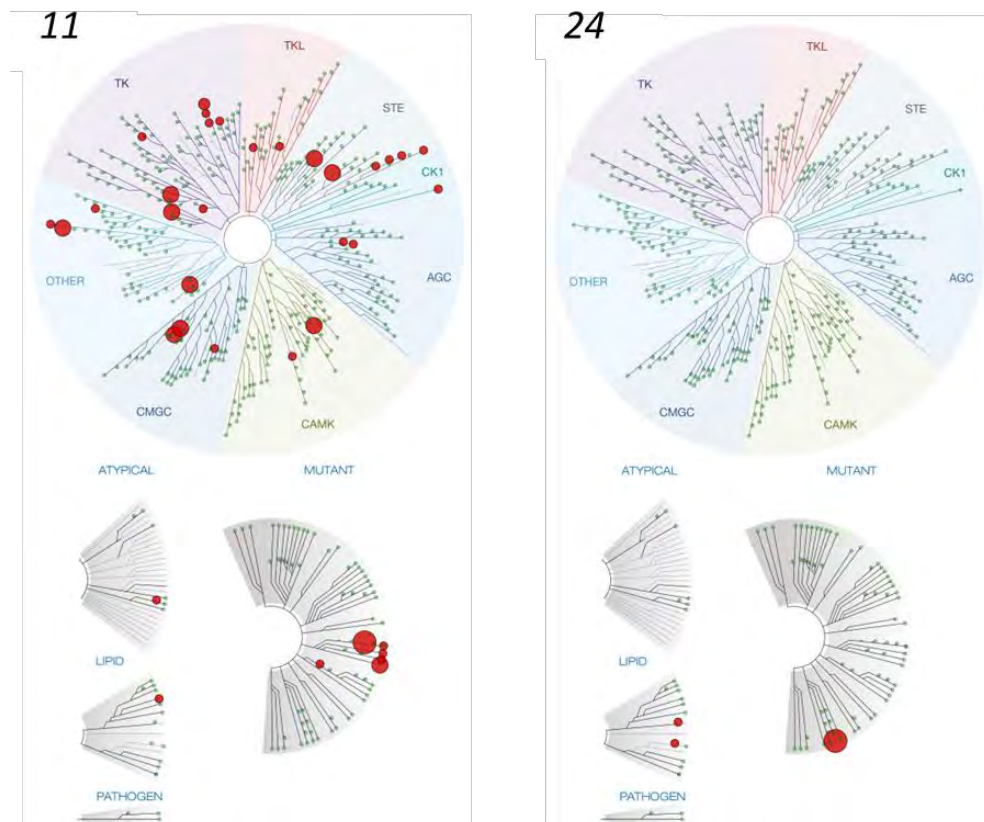
Exploration of the terminal aryl ring revealed several trends (Table 2): (1) Substitution at the 3-position was preferred to substitution at the 2-position of the ring (compare **24** vs. **25**, **26** vs. **27**, **28** vs. **29**, and **30** vs. **31**). (2) Lipophilic substituents, such as CF<sub>3</sub>, Cl, Br, CCH, CF<sub>2</sub>H, and *c*-Pr provided enhanced biochemical and cellular potency compared with F substitution (**24** vs. **26**, **28**, **33**, **32**, **34**, and **35**) leading to the recovery of the potency lost with the incorporation of the selectivity-enhancing 2-methyl pyrrolopyrimidine ring (compare **11** vs. **24** and **11** vs. **26**, **28**, **33**, and **32**). (3) Polar substituents were not well tolerated and resulted in substantially reduced cellular potency, as in the case of CN derivative **36**, or in complete loss of activity (biochemical and cellular activity at the concentrations tested), as in the case of sulfone analog **37**. (4) 3,5-Di-substitution with lipophilic groups provided further enhancements in potency (**38**, **39**, **40**, and **41**). Interestingly, addition of a fluorine substituent to CN on the terminal phenyl ring largely reversed the loss of cellular potency due to the CN group (**36** vs. **42**). (5) Replacement of the terminal phenyl ring with heterocycles, such as F-, Cl-, or CF<sub>3</sub>-substituted pyridines, resulted in 8- to 11-fold losses in cellular potency (compare **24** vs. **43**, **28** vs. **44**, and **26** vs. **45**).





**Figure 1.** Crystal structures of PERK with selected inhibitors **11** (A) and **24** (B). Overlay of the crystal structures of **11** and **24** (C). The same orientation is presented for all panels. The PERK backbone is shown as a ribbon or solid strand in yellow for **11** and coral for **24**. The side chains are depicted as cylinders in the same color as the main chain for carbon, blue for nitrogen, and red for oxygen atoms. Inhibitors are shown as cylinders with same color scheme except green for carbon in panels (A,B) and grey for fluorine. Hydrogen bonds are shown as cyan dashed lines.

While having a minor effect on PERK potency, the methyl substituent on the pyrrolopyrimidine ring had a profound effect on the kinase selectivity. Compounds **11** and **24** were screened in the DiscoverX scanMAX<sup>SM</sup> panel of 468 kinases [26]. The resulting TREEspot<sup>TM</sup> interaction maps (Figure 2) revealed that the methyl substituent in **24** led to superior selectivity versus **11**. The S(35), S(10), and S(1) selectivity scores at a concentration of 1  $\mu$ M were 0.005, 0, and 0 for **24** versus 0.072, 0.035, and 0 for **11**. Given the superior selectivity of **24** but slight loss in potency versus **11**, further optimization of the series was carried out with the 2-methyl substitution on the pyrrolopyrimidine ring to further im-



**Figure 2.** The TREEspot™ interaction map of **11** and **24** assayed at 1000 nM in the 468-kinase scanMAXSM panel. The panel includes 403 unique wild-type kinases and 59 mutant isoforms. Note that PERK is not included in the TREEspot assay but was evaluated in separate biochemical assays. Complete tabulated results presented in Supplemental Table S2. Complete tabulated results presented in Supplemental Table S2.

Exploration of the terminal aryl ring revealed several trends (Table 2): (1) Substitution at the 3-position was preferred to substitution at the 2-position of the terminal aryl ring (compare **24** vs. **25**, **26** vs. **27**, **28** vs. **29**, and **30** vs. **31**). (2) Lipophilic substituents, such as CF<sub>3</sub>, Cl, Br, C(CH<sub>3</sub>)<sub>2</sub>H, and *c*-Pr, provided enhanced biochemical and cellular potency compared with F substitution (**24** vs. **26**, **28**, **33**, **32**, **34**, and **35**) leading to the recovery of the potency lost with the incorporation of the selectivity-enhancing 2-methylpropanoic acid ring (compare **11** vs. **24** and **11** vs. **26**, **28**, **33**, and **34**, **52**) *in vitro* and *in vivo* cellular potency (8- to 36-fold) in the case of (**24** vs. **61** and **60**). (3) Addition of substitution with lipophilic groups provided further enhancements in potency (**38**, **39**, **40**, and **41**). Interestingly, reduction of bioactivity was observed for *in vivo* cellular potency (8- to 36-fold) in the case of (**24** vs. **62** and **63**). (4) Addition of substitution with lipophilic groups provided further enhancements in potency (**38**, **39**, **40**, and **41**). Interestingly, reduction of bioactivity was observed for *in vivo* cellular potency (8- to 36-fold) in the case of (**24** vs. **62** and **63**). (5) Replacement of the terminal phenyl ring with heterocycles (**67**, **68**, **69**, **70**, **71**, **72**, **73**, **74**, **75**, **76**, **77**, **78**, **79**, **80**, **81**, **82**, **83**, **84**, **85**, **86**, **87**, **88**, **89**, **90**, **91**, **92**, **93**, **94**, **95**, **96**, **97**, **98**, **99**, **100**, **101**, **102**, **103**, **104**, **105**, **106**, **107**, **108**, **109**, **110**, **111**, **112**, **113**, **114**, **115**, **116**, **117**, **118**, **119**, **120**, **121**, **122**, **123**, **124**, **125**, **126**, **127**, **128**, **129**, **130**, **131**, **132**, **133**, **134**, **135**, **136**, **137**, **138**, **139**, **140**, **141**, **142**, **143**, **144**, **145**, **146**, **147**, **148**, **149**, **150**, **151**, **152**, **153**, **154**, **155**, **156**, **157**, **158**, **159**, **160**, **161**, **162**, **163**, **164**, **165**, **166**, **167**, **168**, **169**, **170**, **171**, **172**, **173**, 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biochemical data was observed (46 vs. 70 and 71). Replacement of the proximal phenyl ring of 24 with a pyridine ring resulted in modest (2- to 3-fold) losses in cellular potency (24 vs. 72 and 74). Some of the lost potency could be restored by replacing the terminal ring F with CF<sub>3</sub> (73 vs. 72 and 75 vs. 74). Replacement of the proximal phenyl ring of 24 with a pyrimidine ring was not tolerated, resulting in a 30-fold loss in biochemical potency and complete loss in cellular activity (24 vs. 76).  
**Table 3.** In vitro data of PERK inhibitors with proximal and terminal ring modifications.  
**Table 3. Cont.**

Cmpd No.	Chemical Structure	Substituents	hPERK IC <sub>50</sub> (μM)	p-eIF2α IC <sub>50</sub> (μM)
74		3-F, R <sub>2</sub>	0.0083	0.146
75		3-CF <sub>3</sub>	0.0068	0.095
75		3-CF <sub>3</sub> , 3-F	0.0068	0.095
24		3-F, 3-CF <sub>3</sub>	0.020	0.051
766		3-F, 3-F	0.647	>10
76		3-F, 3-F	0.0022	0.041

~Racemic mixture of alcohols.  
 ~Racemic mixture of alcohols.

The SAR campaign successfully identified a number of PERK inhibitors that demonstrated moderate to good oral bioavailability. These promising compounds were evaluated in a rat PK study to determine their Caco-2 permeability (Table 4) in general, the compounds showed moderate kinetic aqueous solubility (Table 4). In general, the in vitro stability/ aqueous solubility, Caco-2 permeability, plasma protein binding, and hepatocyte stability to determine their drug-like properties (Table 4). In general, the (apparent) predicted good oral absorption properties (Table 4). In general, the

tion of 24 (efflux ratio of 5.2), the compounds showed little to no efflux. Plasma protein binding was found to be high across all species tested, a characteristic common for many kinase inhibitors.

**Table 4.** Solubility, permeability, plasma protein binding and hepatocyte clearance properties of selected PERK inhibitors.

Cmpd No.	Kinetic Aqueous Solubility (μM)	Caco-2		Plasma Protein Binding		Hepatocyte Clearance (t <sub>1/2</sub> min)			
		P <sub>app</sub> (A→B) (10 <sup>-6</sup> cm/s)	Efflux Ratio	Human	Rat	Human	Rat	Mouse	Dog
26	19	21	2.2	99.8 ± 0.0	99.8 ± 0.0	194	166	118	130
28	35	23	1.9	99.8 ± 0.1	99.7 ± 0.0	778	27	62	355
65	58	26	1.6	99.6 ± 0.1	98.9 ± 0.0	390	22	64	567
24	76	9	5.2	99.2 ± 0.2	99.1 ± 0.2	733	23	31	>1000
41	24	20	2.5	99.6 ± 0.2	99.5 ± 0.2	68	16	24	232
39	10	16	2.0	99.9 ± 0.0	99.9 ± 0.0	277	252	320	223
38	19	19	1.9	99.0 ± 0.1	99.1 ± 0.3	897	121	805	460
48	14	24	1.6	99.9 ± 0.0	99.6 ± 0.0	556	25	53	215
66	61	24	1.5	99.3 ± 0.2	98.6 ± 0.1	358	13	167	39
70	66	29	1.7	99.4 ± 0.0	99.3 ± 0.1	360	15	45	268

In vitro metabolic stability was assessed in human, rat, mouse, and dog hepatocytes. In general, the compounds were found to be relatively stable in human and dog hepatocytes and exhibited long half-lives. For several compounds (28, 65, 24, 41, 66, and 70), stability was reduced in rodent hepatocytes, particularly rat. Compounds with alkyl groups (i.e., Me) on the terminal phenyl ring tended to have lower stability (41 vs. 39). Several compounds (26, 39, and 38) maintained good metabolic stability across all species tested.



To connect the in vitro ADME results with the in vivo setting, select compounds (**26**, **28**, **65**, **24**, **41**, **39**, **38**, **48**, **66**, and **70**) were examined in mouse PK studies (Table 5). All of the compounds showed robust oral exposure with  $C_{\max}$  values for most compounds >8000 ng/mL and  $AUC_{0-\text{last}}$  values >25,000 h·ng/mL. The compounds also showed good in vivo half-lives of 2–3.5 h and excellent bioavailability ( $\geq 70\%$ ) apart from compound **41**. Compound **26** stood out with a particularly attractive pharmacokinetics profile with  $C_{\max}$  of 17,200 ng/mL,  $AUC_{0-\text{last}}$  of 126,019 h·ng/mL, half-life of 3.5 h, and bioavailability of 82%, a significant improvement over our earlier analog **24** and was therefore chosen as one of the key leads in this series PERK inhibitors.

**Table 5.** In vivo pharmacokinetics (PK) properties of selected PERK inhibitors in mice. Compounds administered at 10 mg/kg by oral gavage (PO).

Compound	$C_{\max}$ (ng/mL)	$AUC_{0-\text{last}}$ (h·ng/mL)	$t_{1/2}$ (h)	F (%)
<b>26</b>	17,200	126,019	3.5	82
<b>28</b>	14,067	65,764	2.4	84
<b>65</b>	8150	40,105	2.2	129
<b>24</b>	13,100	37,495	2.1	77
<b>41</b>	12,467	26,010	2.7	40
<b>39</b>	14,300	86,444	2.9	78
<b>38</b>	8597	28,932	2.0	94
<b>48</b>	3017	10,889	2.1	70
<b>66</b>	9610	39,835	2.0	126
<b>70</b>	9723	32,258	2.0	117

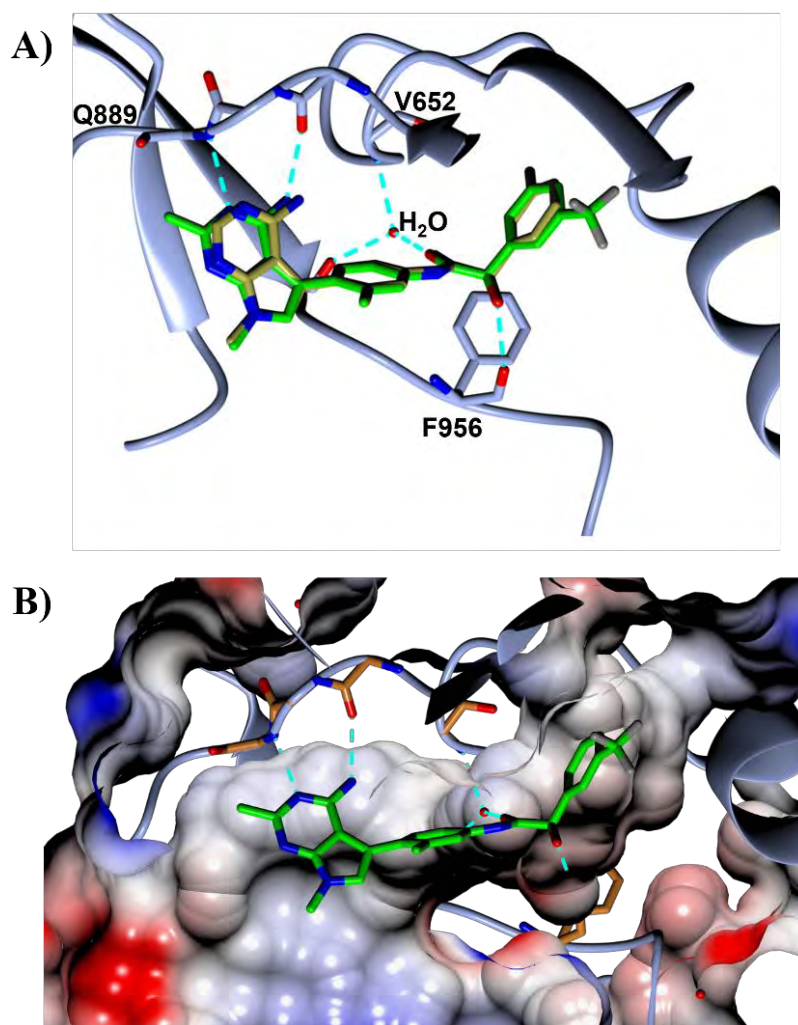
Inhibitor **26** was co-crystallized with PERK into a trigonal space group  $P3_221$  with comparable cell parameters as previously noted with compounds **11** and **24**. Replacement of the 3-fluorine on the terminal phenyl group in **11** and **24** and with a larger  $CF_3$  group in **26** induces a flip in the phenyl ring since the enzyme position containing the fluorine cannot accommodate the bulkier  $CF_3$  group (Figures 1 and 3). The fluorine atom of **24** forms four contacts with a distance of 3.8 Å or less with residues within the back pocket region of PERK, whereas the  $CF_3$  group of **26** forms ten contacts with PERK. These extra contacts may stabilize the binding of **26** to PERK and explain the increased potency of **26** compared to **24**.

Compound **26** was found to have excellent selectivity in the DiscoverX kinase panel (Figure 4; Supplemental Table S2) comparable to that of compound **24**. The  $S(35)$ ,  $S(10)$ , and  $S(1)$  selectivity scores at a concentration of 1  $\mu\text{M}$  were 0.01, 0.002, and 0.002. Importantly, **26** demonstrated potent selectivity for PERK over the other three highly conserved ISR kinases (GCN2, HRI, and PKR; Supplemental Table S3). In biochemical assays, the  $IC_{50}$  of **26** against PERK was 4.7 nM, compared to  $IC_{50}$  values greater than 10  $\mu\text{M}$  for each of the other ISR kinases. The primary off-target activity of **26** was associated with FLT3 kinase and its mutant isoforms. Radiometric kinase assays were used to validate the kinome panel and determine the relative  $IC_{50}$  value of **26** against wild-type FLT3 and a mutant isoform (D835Y). When evaluated individually using concentrations of **26** up to 10  $\mu\text{M}$ , the relative  $IC_{50}$  against wild-type FLT3 was >1  $\mu\text{M}$ , and >10  $\mu\text{M}$  against the mutant variant FLT3 (D385Y), indicating weak to no activity against FLT3wt or the mutant form (Supplemental Table S3).

To understand the effect of **26** in target organs, the PK/PD relationship of **26** was investigated in mouse pancreas. PK analysis of plasma and pancreas samples from mice treated with **26** revealed dose-proportionate exposure at doses ranging from 0.3 to 30 mg/kg following a single administration by oral gavage (Supplemental Figure S2). Pancreas tissue was sampled to evaluate the effect of **26** on PERK autophosphorylation in pancreas, following oral administration of **26** at 0.3, 1, 3, 10, and 30 mg/kg. At 1 h following administration, phosphoPERK (T980), relative to total PERK protein (pPERK/PERK) decreased in a dose-dependent manner, reaching approximately 80% inhibition at 30 mg/kg (Figure 5A). When PERK inhibition was evaluated across time in mouse pancreas, a dose-dependent

48	3017	10,889	2.1	70
66	9610	39,835	2.0	126
70	9723	32,258	2.0	117

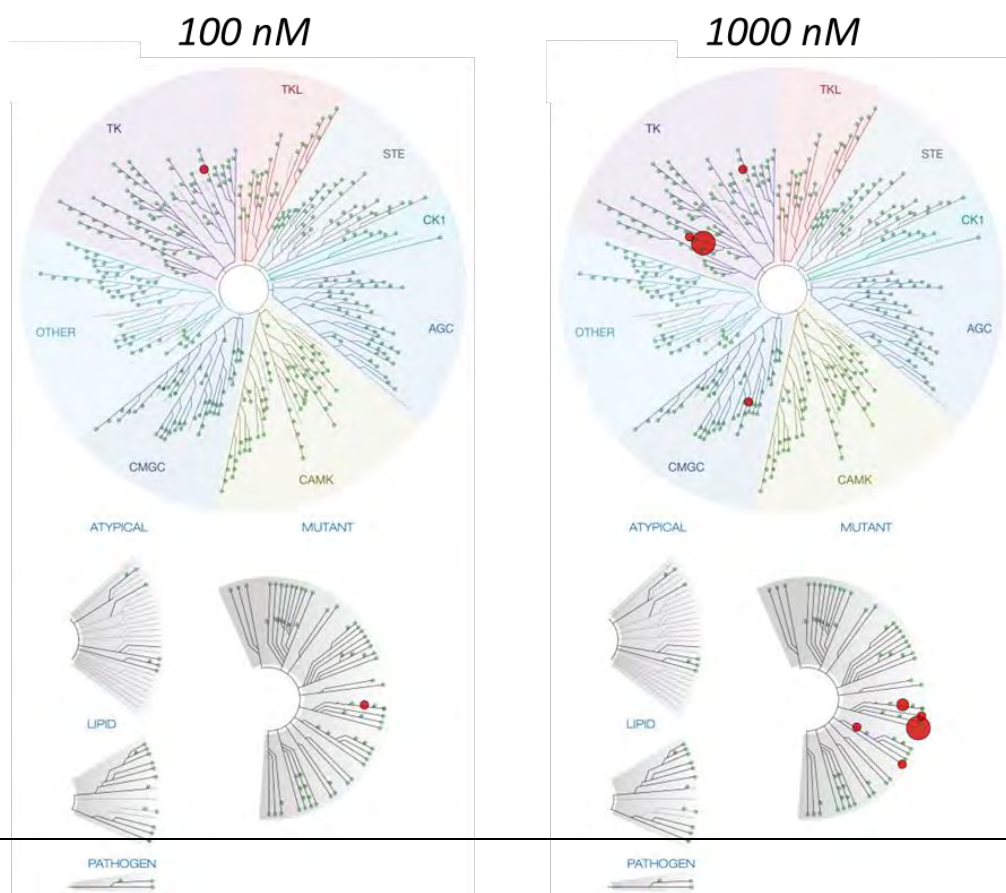
Inhibitor **26** was co-crystallized with PERK into a trigonal space group P3<sub>2</sub>21 with comparable cell parameters as previously noted with compounds **11** and **24**. Replacement of the 3-fluorine on the terminal phenyl group in **11** and **24** and with a larger CF<sub>3</sub> group in **26** induces a flip in the phenyl ring, since the enzyme active site containing the fluorine cannot accommodate the bulkier **26** observed (Figure 4 and 13). The diphenyl atom of **26** forms four contacts with a distance of 3.8 Å or less with residues within the back pocket region of PERK, whereas the CF<sub>3</sub> group of **26** forms ten contacts with PERK. These extra contacts may stabilize the binding of **26** to PERK and explain the increased potency of **26** compared to **24**. In mice, rat and dog both shared similar clearance rates as mouse, albeit with different exposure levels following administration (Figure 5C).



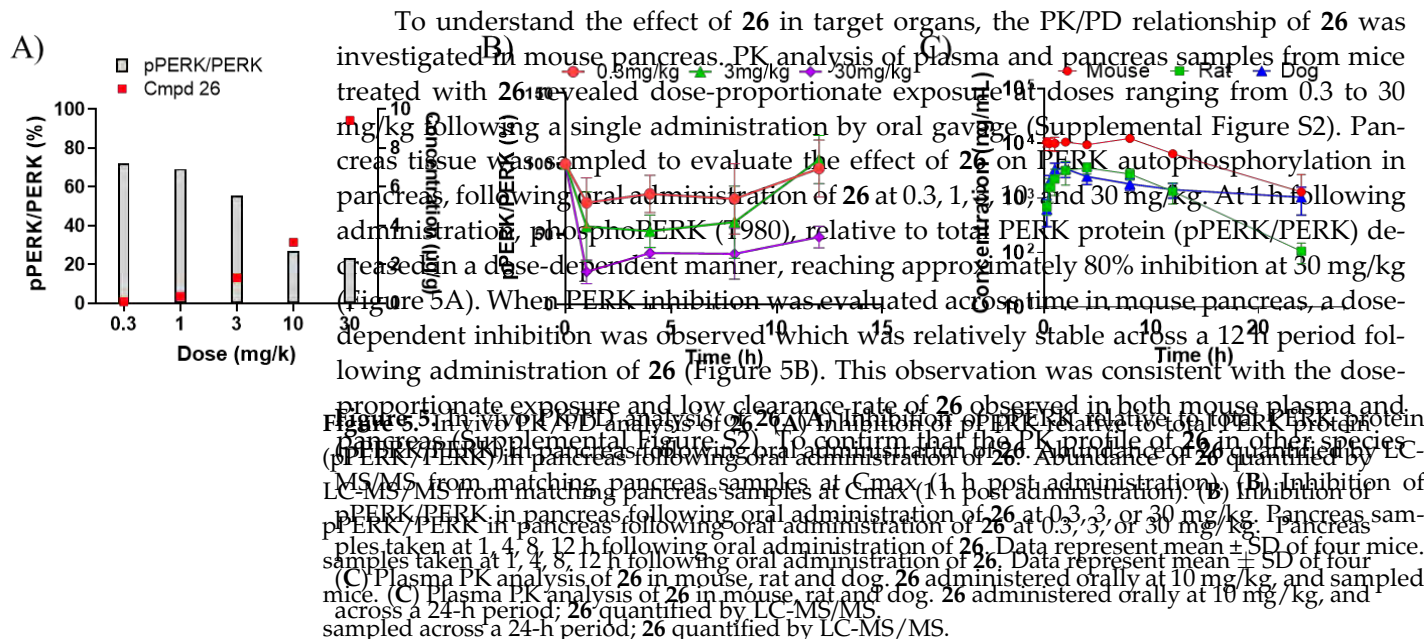
**Figure 3.** Crystal structures of compound **26** bound to the PERK kinase domain. (A) View of **26** in the PERK active site. PERK backbone is shown as a blue ribbon and the side chains are shown as cylinders in brown for carbon, blue for nitrogen, and red for oxygen atoms, respectively. Compound **26** is shown as cylinder with same color scheme except green for carbon and grey for fluorine. Compound **11** is shown in a similar manner to compound **26**, except the color used for carbon atoms. Hydrogen bonds for **26** are shown as cyan dashed lines. The phenyl ring is flipped due to replacement of the 3-fluorine in **11** by CF<sub>3</sub> in **26**. (B) View of **26** in the PERK active site with surface rendering. The color intensity corresponds to the calculated electrostatic potential from -20 kT/e (red) to +20 kT/e (blue).

Compound **26** was found to have excellent selectivity in the DiscoverX kinase panel (Figure 4; Supplemental Table S2) comparable to that of compound **24**. The S(35), S(1), and S(1) selectivity scores at a concentration of 1 μM were 0.01, 0.002, and 0.002. Importantly, **26** demonstrated potent selectivity for PERK over the other th

was available to enable further development, plasma concentration of **26** was quantified in Figure 4. The TRERespot™ interaction map of compound **26** in the 468-kinase scanMAXSM panel at 100 and 1000 nM. The panel includes 403 unique wild-type kinases and 59 mutant isoforms. Note that PERK is not included in the TRERespot assay. Complete tabulated results presented in Supplemental Table S2.



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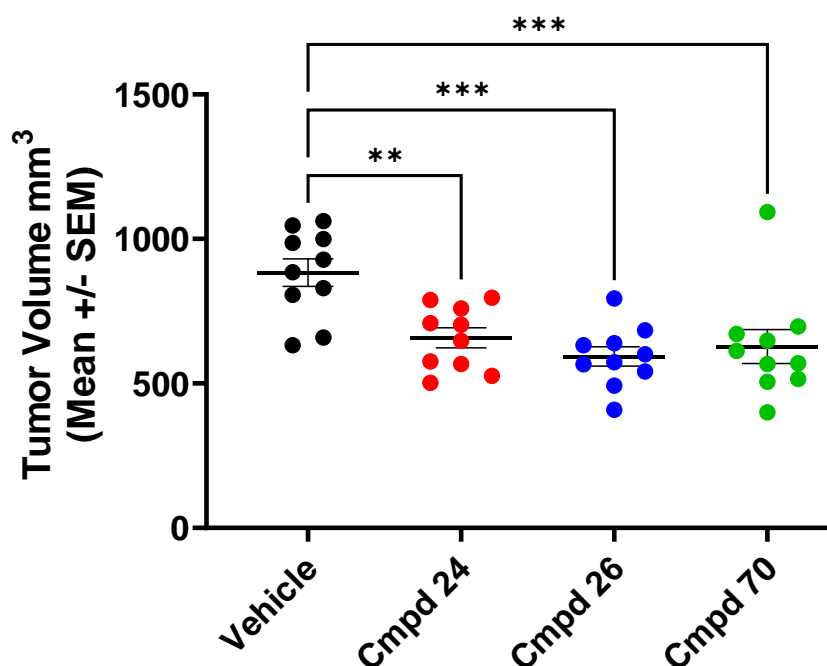
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As the focus of this program was to develop a novel, highly potent, selective and orally bioavailable PERK inhibitor to probe the role of PERK biology in animal models, compounds **24**, **26** and **70** were tested for in vivo efficacy in a clear cell renal cell carcinoma (ccRCC) xenograft model, 786-O (Figure 6). The 786-O model is driven by the Von Hippel-Lindau (VHL) mutation, which leads to PERK pathway activation and is therefore a suitable model to assess the biological activity of these PERK inhibitors [17]. The compounds were dosed at 30 mg/kg, twice per day (BID) for 29 days and were well tolerated as deter-



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**Figure 6.** PERK inhibitors 24, 26 and 70 inhibit tumor growth of 786-O RCC xenografts. Animals were treated at 30 mg/kg BID dosing for 29 days. Statistical analysis of group means by one-way ANOVA (\*\*  $p < 0.01$ , \*\*\*  $p < 0.001$ ).

In summary, we have developed a novel class of potent pyrrolopyrimidine PERK inhibitors with excellent ADME and PK/PD properties. Robust kinase selectivity was achieved with substitution on the pyrrolopyrimidine hinge binder region driving kinase selectivity, with a methyl substitution at the 2-position of the ring resulting in exquisite selectivity for PERK over other kinases. Further optimization of the 2-methyl substituted analogs led to identification of lead molecule **26** that retained excellent potency and selectivity and had good in vitro metabolic stability across all species tested. When **26** was evaluated in vivo, high  $AUC_{last}$  values and a flat, sustained plasma exposure profile were observed in mouse plasma and pancreas tissue. A single administration of **26** resulted in stable PERK inhibition across a 12 h period in mouse pancreas. These characteristics led to the selection of **26** for further development, and evaluation in animal models is ongoing.

**Supplementary Materials:** The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/pharmaceutics14102233/s1>, Supplemental Table S1: Crystallography data collection and refinement statistics; Supplemental Table S2: scanMAXSM Kinome Assay Results; Supplemental Table S3: Potency of Cmpd **26** against four ISR kinases and two FLT3 isoforms; Supplemental Figure S1: Compound **26** is selective against cell lines driven by FLT3-ITD; Supplemental Figure S2: In vivo PK in plasma and pancreas; Supplemental Figure S3: Compound **26** slows growth of 786-O RCC tumor xenografts; Supplemental Scheme S1: Synthesis of (*R*)-*N*-(4-(4-amino-7-methyl-7*H*-pyrrolo[2,3-*d*]pyrimidin-5-yl)-3-methylphenyl)-2-(3-fluorophenyl)-2-hydroxyacetamide (**11**) and (*S*)-*N*-(4-(4-amino-7-methyl-7*H*-pyrrolo[2,3-*d*]pyrimidin-5-yl)-3-methylphenyl)-2-(3-fluorophenyl)-2-hydroxyacetamide (**12**); Supplemental Scheme S2: Synthesis of (*S*)-*N*-(4-(4-amino-2,7-dimethyl-7*H*-pyrrolo[2,3-*d*]pyrimidin-5-yl)-2,3-difluorophenyl)-2-(3-fluorophenyl)-2-hydroxyacetamide (**96**) and (*R*)-*N*-(4-(4-amino-2,7-dimethyl-7*H*-pyrrolo[2,3-*d*]pyrimidin-5-yl)-2,3-difluorophenyl)-2-(3-fluorophenyl)-2-hydroxyacetamide (**70**); Supplemental Scheme S3: Synthesis of (*S*)-*N*-(4-(4-amino-2,7-dimethyl-7*H*-pyrrolo[2,3-*d*]pyrimidin-5-yl)-3-fluorophenyl)-2-hydroxy-2-(3-(trifluoromethyl)phenyl)acetamide (**106**) and (*R*)-*N*-(4-(4-amino-2,7-dimethyl-7*H*-pyrrolo[2,3-*d*]pyrimidin-5-yl)-3-fluorophenyl)-2-hydroxy-2-(3-(trifluoromethyl)phenyl)acetamide (**47**). References [7,17,20–25,27] are cited in the supplementary materials.

**Author Contributions:** Conceptualization, M.J.M., A.C.R., M.E.S. and M.D.S.; methodology, M.J.M., M.E.S., J.G., M.B., V.C., D.S., A.-H.L. and M.D.S.; formal analysis, M.D.S., M.E.S., D.S. and V.C.; investigation, M.E.S., V.C., G.Z., M.B. and J.G.; resources, A.C.R. and M.J.M.; data curation, G.Z. and M.D.S.; writing—original draft preparation, M.J.M., M.E.S., M.D.S. and V.C.; writing—review and editing, M.E.S., M.J.M. and M.D.S.; visualization, H.D. and G.Z.; supervision, A.C.R., M.J.M., M.E.S., J.G., M.B. and M.D.S.; funding acquisition, A.C.R. and M.J.M. All authors have read and agreed to the published version of the manuscript.

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**Institutional Review Board Statement:** All the procedures related to animal handling, care, and the treatment in this study were performed according to guidelines approved by the Institutional Animal Care and Use Committee (IACUC) of Pharmaron following the guidance of the Association for Assessment and Accreditation of Laboratory Animal Care (AAALAC). All animal studies were approved and carried out in accordance with Pharmaron's Institutional Animal Care and Use Committee, protocol ON-TOL-09012021 approved on 26 March 2021.

**Data Availability Statement:** Coordinates and structural factors have been deposited in the PDB under codes 8EQ9 (Compound **11**), 8EQD (Compound **24**), and 8EQE (Compound **26**).

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**Conflicts of Interest:** At the time of execution, M.E.S., V.C., D.S., A.-H.L., A.C.R. and M.J.M. were employed by HiberCell Inc. As such, HiberCell was involved in all aspects of this work, including conceptualization and methodology, investigation, data analysis, resources, and writing of this manuscript. Curia Global is a contract research organization engaged by HiberCell Inc to support chemical synthesis and characterization of novel molecules. At the time of experimentation, M.D.S., M.B., J.G. and G.Z. were employed by Curia Global, which was involved in conceptualization, methodology, resources, data analysis and manuscript composition. Pharmaron Co. Ltd. is a contract research organization engaged by HiberCell to support chemical characterization of novel compounds. At the time of experimentation, H.D. was employed by Pharmaron, which was involved in data visualization and manuscript composition.

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